Propose a mechanism to account for the following conversion of testoterone to the alkene shown:

$$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \\$$

Two compounds (A and B) have formulas C₃H₈O. They gave the following results with three chemical tests. Propose structures for A and B consistent with this information:

- P	Na Metal	Chromic Acid Na ₂ Cr ₄ O ₇ /H ₂ SO ₄	Lucas Reagent ZnCl ₂ /HCl	76. 754
Compound A	Bubbles	Orange to green	15 minutes	26
Compound B	No reaction	No reaction	No reaction	

Arrange the following compounds in increasing order of boiling point. Justify your order in terms of intermolecular forces.