I have these questions:

1. Which one is a weak monoprotic acid: benzoic acid, perchloric acid, oxalic acid , sulfurous acid?
2. What is the pOH of a 0.0130 M aqueous pyridine solution? Kb C5H5N = 1.4 x 10-9
3. What is the pH of a 0.150 M aqueous solution of ammonia? Kb NH3 = 1.8 x 10-5
4. What is the hydronium ion concentration in a 0.100 M hypochlorous acid solution? Ka HClO = 3.5 x 10-8