

**TABLE 1.5**  
Approximate Physical Properties of Some Common Liquids (BG Units)

Liquid	Temperature (°F) †	Density, $\rho$ (slugs/ft <sup>3</sup> )	Specific Weight, $\gamma$ (lb/ft <sup>3</sup> )	Dynamic Viscosity, $\mu$ (lb · s/ft <sup>2</sup> )	Kinematic Viscosity, $\nu$ (ft <sup>2</sup> /s)	Surface Tension, <sup>a</sup> $\sigma$ (lb/ft)	Vapor Pressure, $P_v$ (lb/in. <sup>2</sup> abs)]	Bulk Modulus, <sup>b</sup> $E_v$ (lb/in. <sup>2</sup> )
Carbon tetrachloride	68	3.09	99.5	2.00 E - 5	6.47 E - 6	1.84 E - 3	1.9 E + 0	1.91 E + 5
Ethyl alcohol	68	1.53	49.3	2.49 E - 5	1.63 E - 5	1.56 E - 3	8.5 E - 1	1.54 E + 5
Gasoline <sup>c</sup>	60	1.32	42.5	6.5 E - 6	4.9 E - 6	1.5 E - 3	8.0 E + 0	1.9 E + 5
Glycerin	68	2.44	78.6	3.13 E - 2	1.28 E - 2	4.34 E - 3	2.0 E - 6	6.56 E + 5
Mercury	68	26.3	847	3.28 E - 5	1.25 E - 6	3.19 E - 2	2.3 E - 5	4.14 E + 6
SAE 30 oil <sup>c</sup>	60	1.77	57.0	8.0 E - 3	4.5 E - 3	2.5 E - 3	—	2.2 E + 5
Seawater	60	1.99	64.0	2.51 E - 5	1.26 E - 5	5.03 E - 3	2.26 E - 1	3.39 E + 5
Water	60	1.94	62.4	2.34 E - 5	1.21 E - 5	5.03 E - 3	2.26 E - 1	3.12 E + 5

<sup>†</sup>In contact with air.

<sup>b</sup>Isentropic bulk modulus calculated from speed of sound.

<sup>c</sup>Typical values. Properties of petroleum products vary.

**TABLE 1.6**  
Approximate Physical Properties of Some Common Liquids (SI Units)

Liquid	Temperature (°C)	Density, $\rho$ (kg/m <sup>3</sup> )	Specific Weight, $\gamma$ (kN/m <sup>3</sup> )	Dynamic Viscosity, $\mu$ (N · s/m <sup>2</sup> )	Kinematic Viscosity, $\nu$ (m <sup>2</sup> /s)	Surface Tension, <sup>a</sup> $\sigma$ (N/m)	Vapor Pressure, $P_v$ (N/m <sup>2</sup> abs)]	Bulk Modulus, <sup>b</sup> $E_v$ (N/m <sup>2</sup> )
Carbon tetrachloride	20	1,590	15.6	9.58 E - 4	6.03 E - 7	2.69 E - 2	1.3 E + 4	1.31 E + 9
Ethyl alcohol	20	789	7.74	1.19 E - 3	1.51 E - 6	2.28 E - 2	5.9 E + 3	1.06 E + 9
Gasoline <sup>c</sup>	15.6	680	6.67	3.1 E - 4	4.6 E - 7	2.2 E - 2	5.5 E + 4	1.3 E + 9
Glycerin	20	1,260	12.4	1.50 E + 0	1.19 E - 3	6.33 E - 2	1.4 E - 2	4.52 E + 9
Mercury	20	13,600	133	1.57 E - 3	1.15 E - 7	4.66 E - 1	1.6 E - 1	2.85 E + 10
SAE 30 oil <sup>c</sup>	15.6	912	8.95	3.8 E - 1	4.2 E - 4	3.6 E - 2	—	1.5 E + 9
Seawater	15.6	1,030	10.1	1.20 E - 3	1.17 E - 6	7.34 E - 2	1.77 E + 3	2.34 E + 9
Water	15.6	999	9.80	1.12 E - 3	1.12 E - 6	7.34 E - 2	1.77 E + 3	2.15 E + 9

<sup>a</sup>In contact with air.

<sup>b</sup>Isentropic bulk modulus calculated from speed of sound.

<sup>c</sup>Typical values. Properties of petroleum products vary.

■ TABLE 1.7  
Approximate Physical Properties of Some Common Gases at Standard Atmospheric Pressure (BG Units)

Gas	Temperature (°F)	Density, $\rho$ (slugs/ft <sup>3</sup> )	Specific Weight, $\gamma$ (lb/ft <sup>3</sup> )	Dynamic Viscosity, $\mu$ (lb · s/ft <sup>2</sup> )	Kinematic Viscosity, $\nu$ (ft <sup>2</sup> /s)	Gas Constant, <sup>a</sup> $R$ (ft · lb/slug · °R)	Specific Heat Ratio, <sup>b</sup> $k$
Air (standard)	59	2.38 E - 3	7.65 E - 2	3.74 E - 7	1.57 E - 4	1.716 E + 3	1.40
Carbon dioxide	68	3.55 E - 3	1.14 E - 1	3.07 E - 7	8.65 E - 5	1.130 E + 3	1.30
Helium	68	3.23 E - 4	1.04 E - 2	4.09 E - 7	1.27 E - 3	1.242 E + 4	1.66
Hydrogen	68	1.63 E - 4	5.25 E - 3	1.85 E - 7	1.13 E - 3	2.466 E + 4	1.41
Methane (natural gas)	68	1.29 E - 3	4.15 E - 2	2.29 E - 7	1.78 E - 4	3.099 E + 3	1.31
Nitrogen	68	2.26 E - 3	7.28 E - 2	3.68 E - 7	1.63 E - 4	1.775 E + 3	1.40
Oxygen	68	2.58 E - 3	8.31 E - 2	4.25 E - 7	1.65 E - 4	1.554 E + 3	1.40

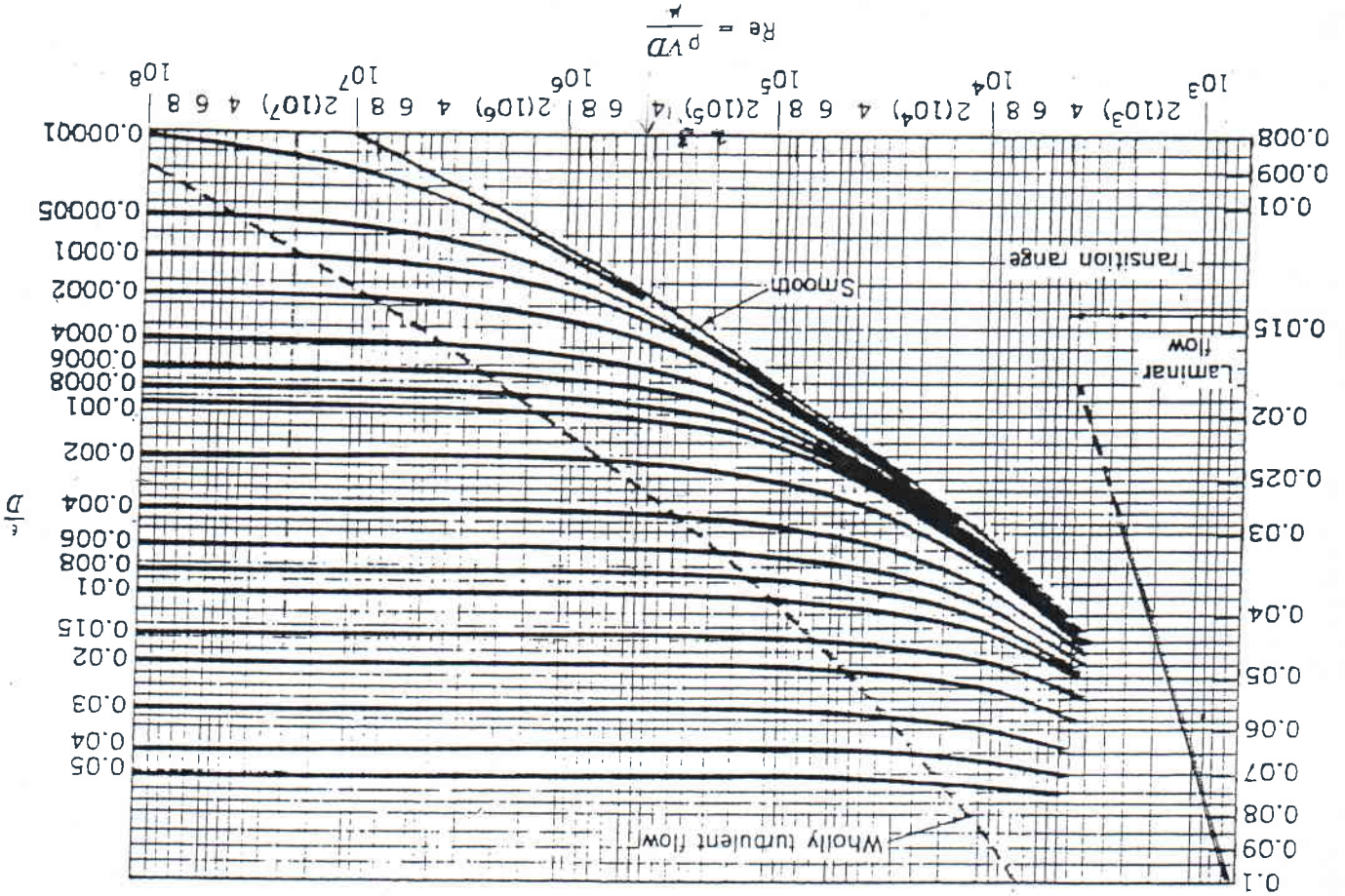
<sup>a</sup>Values of the gas constant are independent of temperature.  
<sup>b</sup>Values of the specific heat ratio depend only slightly on temperature.

■ TABLE 1.8  
Approximate Physical Properties of Some Common Gases at Standard Atmospheric Pressure (SI Units)

Gas	Temperature (°C)	Density, $\rho$ (kg/m <sup>3</sup> )	Specific Weight, $\gamma$ (N/m <sup>3</sup> )	Dynamic Viscosity, $\mu$ (N · s/m <sup>2</sup> )	Kinematic Viscosity, $\nu$ (m <sup>2</sup> /s)	Gas Constant, <sup>a</sup> $R$ (J/kg · K)	Specific Heat Ratio, <sup>b</sup> $k$
Air (standard)	15	1.23 E + 0	1.20 E + 1	1.79 E - 5	1.46 E - 5	2.869 E + 2	1.40
Carbon dioxide	20	1.83 E + 0	1.80 E + 1	1.47 E - 5	8.03 E - 6	1.889 E + 2	1.30
Helium	20	1.66 E - 1	1.63 E + 0	1.94 E - 5	1.15 E - 4	2.077 E + 3	1.66
Hydrogen	20	8.38 E - 2	8.22 E - 1	8.84 E - 6	1.05 E - 4	4.124 E + 3	1.41
Methane (natural gas)	20	6.67 E - 1	6.54 E + 0	1.10 E - 5	1.65 E - 5	5.183 E + 2	1.31
Nitrogen	20	1.16 E + 0	1.14 E + 1	1.76 E - 5	1.52 E - 5	2.968 E + 2	1.40
Oxygen	20	1.33 E + 0	1.30 E + 1	2.04 E - 5	1.53 E - 5	2.598 E + 2	1.40

Values of the gas constant are independent of temperature.  
Values of the specific heat ratio depend only slightly on temperature.

FIGURE 8.23 Friction factor as a function of Reynolds number and relative roughness for round pipes—the Moody chart (Data from Ref. 7 with permission).



Pipe	Feet	Millimeters
Riveted steel	0.003-0.03	0.9-9.0
Concrete	0.001-0.01	0.3-3.0
Wood stave	0.0006-0.003	0.18-0.9
Cast iron	0.00085	0.26
Galvanized iron	0.0005	0.15
Commercial steel or wrought iron	0.00015	0.045
Drawn tubing	0.000005	0.0015
Plastic, glass	0.0 (smooth)	0.0 (smooth)

TABLE 8.1 Equivalent Roughness for New Pipes [From Moody (Ref. 7) and Colebrook (Ref. 8)]

Equivalent Roughness,  $\epsilon$

■ TABLE 8.2

Loss Coefficients for Pipe Components ( $h_L = K_L \frac{V^2}{2g}$ ) (Data from Refs. 5, 10, 27)

Component	$K_L$		
<b>a. Elbows</b>			
— Regular 90°, flanged	0.3		
Regular 90°, threaded	1.5		
Long radius 90°, flanged	0.2		
Long radius 90°, threaded	0.7		
Long radius 45°, flanged	0.2		
Regular 45°, threaded	0.4		
<b>b. 180° return bends</b>			
180° return bend, flanged	0.2		
180° return bend, threaded	1.5		
<b>c. Tees</b>			
Line flow, flanged	0.2		
Line flow, threaded	0.9		
Branch flow, flanged	1.0		
Branch flow, threaded	2.0		
<b>d. Union, threaded</b>	0.08		
<b>*e. Valves</b>			
— Globe, fully open	10		
Angle, fully open	2		
Gate, fully open	0.15		
Gate, $\frac{1}{4}$ closed	0.26		
Gate, $\frac{1}{2}$ closed	2.1		
Gate, $\frac{3}{4}$ closed	17		
Swing check, forward flow	2		
Swing check, backward flow	$\infty$		
Ball valve, fully open	0.05		
— Ball valve, $\frac{1}{3}$ closed	5.5		
— Ball valve, $\frac{2}{3}$ closed	210		

\*See Fig. 8.32 for typical valve geometry.



## TURBOMACHINE SIMILAKITY RELATIONS:

(ASSUMING  
HOMOLOGOUS  
CONDITIONS)

h: head

$\omega$ : angular  
velocity

T: TORQUE

P: POWER

D: Diameter

Q: Flow rate

$K_i$ : constants

TURBINES	PUMPS
$\omega = K_1 \frac{\sqrt{h}}{D}$	$h = K_1 D^2 \omega^2$
$Q = K_2 D^2 \sqrt{h}$	$Q = K_2 D^3 \omega$
$T = K_3 \rho D^3 h$	$T = K_3 \rho D^5 \omega^2$
$P = K_4 \rho D^2 h^{3/2}$	$P = K_4 \rho D^5 \omega^3$

## TURBOMACHINE EFFICIENCIES:

VOLUMETRIC,  $\eta_v$

$$\frac{Q - Q_L}{Q}$$

Q: PASSING THRU  
TURBINE

$$\frac{Q}{Q + Q_L}$$

Q: DELIVERED  
BY PUMP

HYDRAULIC,  $\eta_h$

$$\frac{h_T - h_f}{h_T}$$

$h_T$ : from Energy Eqn

$h_f$ : Analogous To pipe loss in Turbine / PUMP

$$\frac{h_p}{h_p + h_f}$$

$h_p$ : from Energy Eqn

MECHANICAL,  $\eta_m$

$$\frac{b_p}{b_p + f_p} = \frac{T \omega}{(T + T_f) \omega}$$

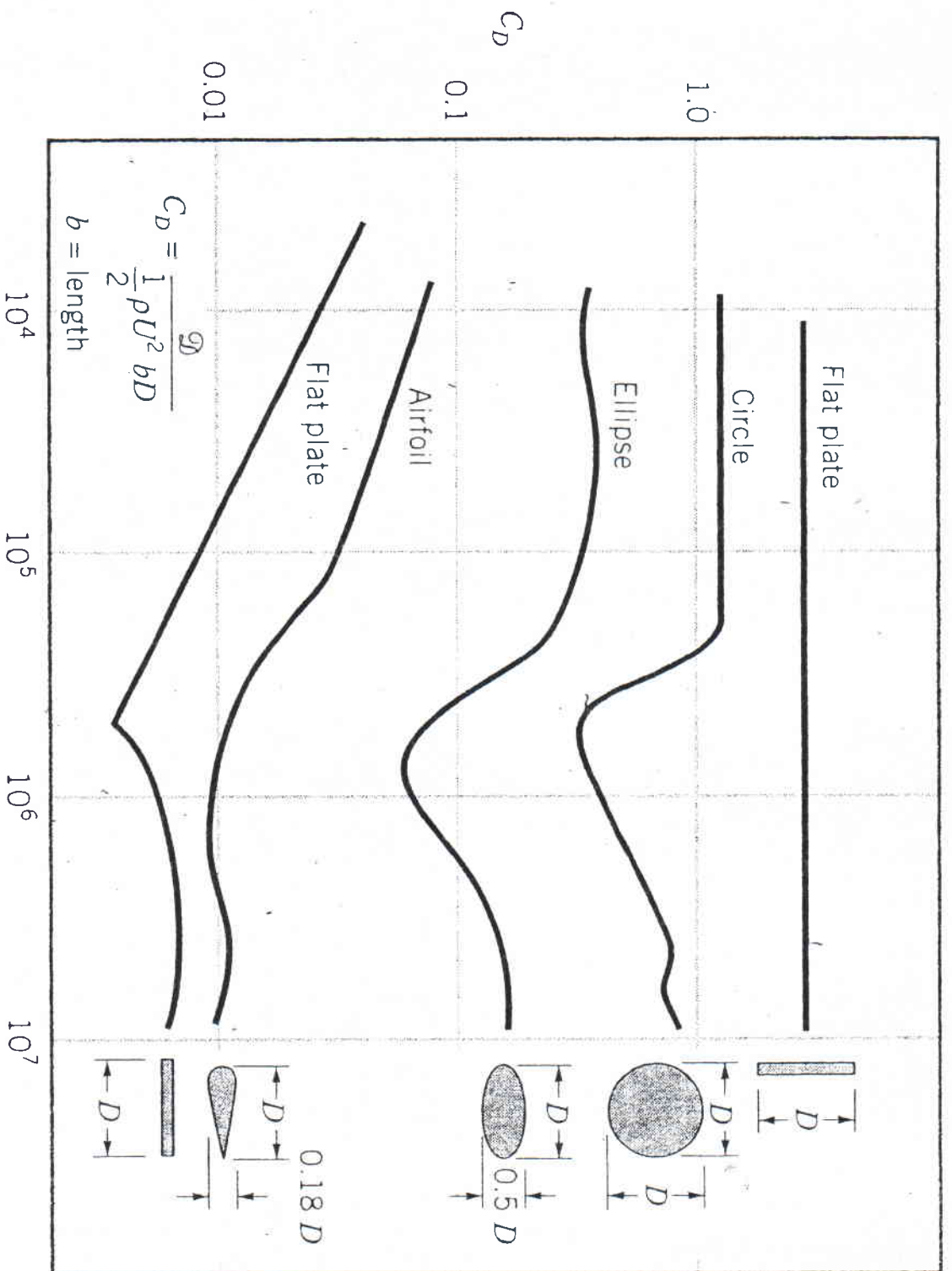
$$\frac{b_p - f_p}{b_p}$$

OVERALL,  $\eta$

$$\frac{T \omega}{\rho Q h_T}$$

$$\frac{\rho Q h_p}{T \omega}$$

$$\eta = \eta_v \cdot \eta_h \cdot \eta_m$$



■ **FIGURE 9.22** Character of the drag coefficient as a function of Reynolds number for objects with various degrees of streamlining, from a flat plate normal to the upstream flow to a flat plate parallel to the flow (two-dimensional flow) (Ref. 5).