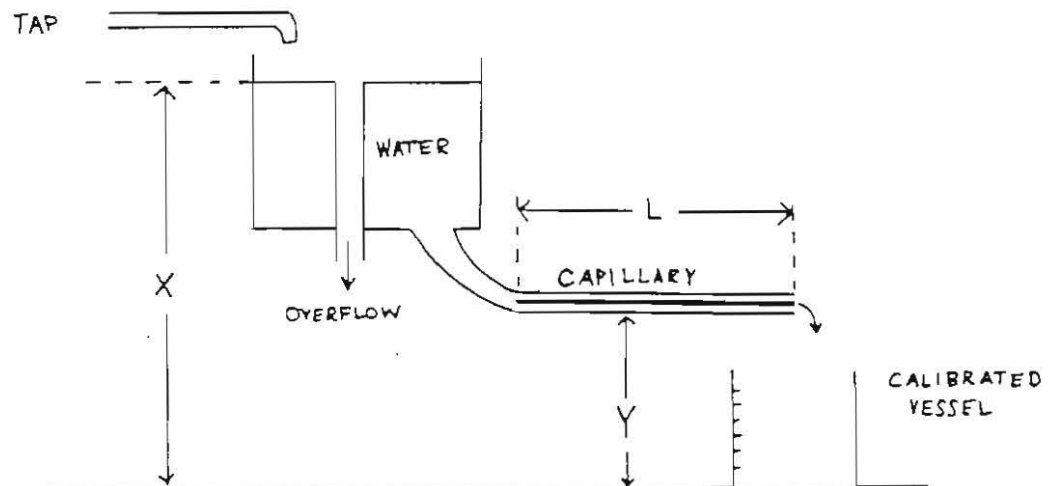


PHYS 2117E Exp. 7 Viscosity of a Liquid

PURPOSE: to measure the viscosity of water by capillary flow method.

- REFERENCES:
- 1) Sears, Zemansky and young, UNIVERSITY PHYSICS, 1982 Edition, Section 13-6.
  - 2) The Physics Teacher, May 1990, pp. 297-299.

APPARATUS: the system components used are shown in the following figure.



THEORY: The rate of flow of a liquid ( $\text{cm}^3/\text{s}$ ) through a capillary is directly proportional to the pressure gradient across the capillary:

$$\frac{dV}{dt} = \frac{\pi}{8} \frac{R^4}{\eta} \frac{(P_1 - P_2)}{L}$$

$$= \frac{\pi}{8} \frac{R^4 \Delta P}{\eta L}$$

where  $V$  is the volume of water

$R$  is the radius of capillary

$\eta$  is the viscosity of liquid

$P_1, P_2$  are the pressures at the ends of the capillary

$L$  is the length of the capillary

Note how sensitively the rate of flow depends on the radius of the capillary. As a result for instance, if the human heart is to provide a steady supply of blood to the brain, a 10% reduction in the radius of the capillaries carrying the blood would require the blood pressure to increase by 50% to maintain this supply of blood.

The above equation is known as Poiseuille's Law, which is applied in LAMINAR flow.

Since the rate of flow  $\frac{dV}{dt} = dQ$  is linearly proportional to the

pressure difference  $\Delta P$ , we have  $\frac{dQ}{d(\Delta P)} = S = \frac{\pi R^4}{8\eta L}$  and the

viscosity  $\eta$  can be obtained from  $\eta = \frac{\pi R^4}{8SL}$ . It is usual to

measure R, and L in cm and Q in  $\frac{cm^3}{s}$  and  $\Delta P$  in dynes to obtain

$$\eta \text{ in poises} = \frac{g}{cms}$$

PROCEDURE: In the experiment, a reservoir of water, A, with a constant level  $x$  cm above the bench feeds into a capillary of length L which must be set perfectly horizontal using a level. The capillary discharges into a cylinder C with graduations on it.

One measures the rate of flow through the capillary by measuring, with a stop watch, the time it takes until the cylinder C is filled up to a certain mark. This measurement should be repeated 5 times. The pressure difference between both ends of the capillary is expressed in gauge pressure, so it is zero at the open

end, and  $(x-y)pg$  at the other end,  $g = 981 \frac{cm}{s^2}$ , the gravitational

constant,  $\rho$  the density of water ( $= 1g/cm^3$ ).

The measurements are repeated at least at 5 different values for  $y$ , the height of the capillary above the bench, in order to obtain the rate of flow of water at different pressures.  $x$  will be constant.

One then plots  $\frac{dV}{dt} = dQ$  vs. the pressure  $(x-y)gp$  and calculates

the slope S from a best-fit straight line:

Note: viscosity depends on temperature. The temperature of tap water may fluctuate. Start the water flow - through the waste drain - 15 minutes before doing the experiment and keep the thermometer in the reservoir. Record and list the temperature with your result.

The experiment should be repeated for as many capillaries as possible. The requirement of LAMINAR flow restricts the dimensions of capillaries to specific values. A detailed discussion of this can be found in Ref. 2.